

Research Article

Simultaneous Determination of Albendazole and Praziquantel Using Zero Order Spectrophotometry Absorbance correction Methods in Veterinary Pharmaceutical Formulation

Suddhasattya Dey¹, Robina Khatun¹, Rajiv Jash¹, Shreya Shah², Pankaj Dagur², Padmacharan Behera^{*3}, Sourav Khawas³ and Anirban Nandi⁴

¹Department of Pharmacy, Sanaka Educational Trust's Group of Institutions, West Bengal713212, India

²Sigma Institute of Pharmacy, Bakrol, Vadodara, Gujrat, India

³Jharkhand Rai University, Ratu Rd, Kamre, Ranchi Jharkhand 835222

⁴Bengal College Of Pharmaceutical Technology Dubrajpur, birbhum, West Bengal

ARTICLE INFO ABSTRACT

Date of submission:	The present study describes a simple, accurate, precise and cost
15-06-2022	effective UV-VIS Spectrophotometric method for the estimation
Date of Revision:	of albendazole (ABZ) and praziquantel (PZQ) by absorbance
21-06-2022	correction method. The simultaneous determination of albendazole
Date of	(ABZ) and praziquantel (PZQ) was performed by absorbance correction method using two different wavelengths i.e. 217nm &
acceptance:	295.4nm. Both the drugs were dissolved in methanol for
07-07-2022	estimation. A linear response was observed in the range of 4-
Key Words: Albendazole, Praziquantel, UV- VIS, Absorbance Correction Method, LOD	then validated for different parameters as per the ICH (International Conference for Harmonization) guidelines. The detection limits (LOD = 4 µg/ml for both the drugs) for absorbance correction method were determined and presented the best analytical features. The recoveries of ABZ and PZQ from the synthetic samples were near to $100 \pm 5\%$. The methods were applied in veterinary pharmaceutical formulation whose mass ratio ABZ: PZQ is 12:1; the results obtained were according to nominal content.

©2020 Published by HOMES on behalf of RJPLS This is an open access article under the CC-BY-NC-ND License.

*Corresponding author:

Padmacharan Behera

Jharkhand Rai University, Ratu Rd, Kamre, Ranchi Jharkhand – 835222 E-mail address: <u>padma.behera@jru.edu.in</u>, Tel: +91 7739897317.

1. Introduction

The antiparasitic drugs are widely used in veterinary practice and human therapy. The antihelminthics are an important type of these drugs used in cattle and pets, bringing benefit and improved quality of life to such animals.⁽¹⁾An important example of a disease treated with these drugs is helminthiases, a common parasitic disease of great economical and public importance. Since health the antihelminthic spectra of most drugs used for treatment is limited, combinations of more than one active ingredient are required to control helminthic infections effectively. In this context, albendazole, methyl 5-(propylthio)-2- benzimidazole carbamate (ABZ) (Fig. 1a), is an antihelminthic drug, it is active against most of the nematode worms and some of the cestode worms in humans and animals, and it acts by inhibiting fumarate reductase and the microtubular polymerization of the parasite.⁽²⁾ Praziquantel, 2cyclohexylcarbonyl-

1,2,3,6,7,11bhexahydro- 4h-pyrazino(2,1a) isoquinolin-4-one (PZQ) (Fig. 1b), acts on the parasite, increasing the permeability to calcium ions causing contractions and vacuolization. The mix of both drugs is efficiently used in veterinary treatment of parasitic diseases. Many publications describe the determination of ABZ and PZQ in pharmaceutical formulations. Individual and simultaneous determination with other drugs are possible by FIA using UV-detection and HPLC method in suspensions,⁽³⁾ adsorptive stripping voltammetry (ASV) and linear sweep voltammetry (LSV), square-wave voltammetry (SWV), differential pulse (DPV),⁽⁴⁻⁵⁾ voltammetry Visspectrophotometry,^(6–9) HPLC,^(10,11) derivative UV-spectrophotometry,⁽¹²⁻¹³⁾ GLC⁽¹⁴⁾ and PMR spectrometry.⁽¹⁵⁾ In biological samples for individuals, simultaneous determinations with other drugs and metabolites have been reported by: HPLC, (16-22) LC-TMS, (23-24) LC-ES-SM,⁽²⁵⁾ nonaqueous capillary electrophoresis,⁽²⁶⁾ IR⁽²⁷⁾ and fluorometry.(28-29)

A newer dose regiment was developed in combination of ABZ and PZQ for treatment of helminthes and the estimation was also been performed by RP-HPLC in rat plasma.⁽³⁰⁾ Classical least squares (CLS) analysis is one of the simplest multivariate methods and is easy to perform, although its results are not very accurate in the quantification of mixtures when the analyzed spectra have significant overlapping.⁽³¹⁾Second derivative spectroscopic method was developed for Veterinary the estimation in Pharmaceutical Formulation .⁽³²⁾ On the other hand, PCR and PLS regression have been used successfully more in quantification of those types of samples, although their use presents more complications than CLS and second derivative spectrophotometry (SDS). The aim of this work is to develop different mathematical approaches for the simultaneous determination of ABZ and PZQ in veterinary pharmaceutical formulation.

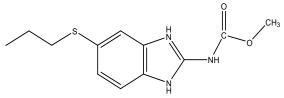


Fig. 1.Chemical structures of Albendazole

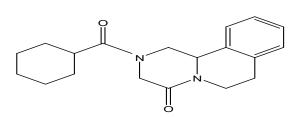


Fig. 2.Chemical structures of Praziquantel

2. Experimental

2.1. Materials and Solutions

Albendazole (Fig. 1) was obtained from Mercury Pharmaceutical Ltd, Vadodara, Gujarat, India. Praziquantel (Fig. 2) was obtained from Micro Labs Ltd., Goa. The solvent used was methanol which was of AR grade, purchased from SD Fine Chemicals Limited, India and double distilled water.

2.2. Instrumentation:

The instrument used for the present study was a UV-Vis double beam spectrophotometer (model 2080, Analytical Technological Limited) with 1cm matched pair quartz cell.

2.3. Method:

2.3.1. Solubility Test:

Solubility test for the drug Albendazole and Praziquantel was performed by using various solvents. The solvents include water, methanol, ethanol, acetonitrile, 0.1N hydrochloric acid (0.1N HCl), 0.1N sodium hydroxide (0.1N NaOH), DMF, and chloroform.

Solvent: Methanol

2.3.2. Preparation of stock solution of Albendazole and Praziquantal (1000 μ g/ml): 25 mg of both the drugs were weighed and transferred to a 25ml volumetric flask. 15 ml of diluent was added and the solution is sonicated for 15 min. Volume is made up to the mark with diluent to obtain stock solution of 1000 μ g/ml.

2.3.3. Preparation of working standard solution Albendazole and Praziquantal (100μg/ml): 2.5 ml of stock solution was withdrawn and transferred to 25 ml volumetric flask, volume is made upto the mark with Solvent to get the working standard solution of 100 μg/ml.

2.3.4. Absorbance Correction Method:

Each working standard solution was scanned between the range 200-400 nm in 1cm cell against blank and the overlain spectra was recorded (Fig.3). Albendazole shows two absorbance maxima (λ max) at 295.4 nm and 215.8 nm wavelengths, whereas Praziquantel shows no absorbance maxima (λ max), but it shows Absorbance at 217.0 nm. (Fig.3). Overlain spectra of Albendazole (ABZ) and Praziquantel (PRQ). The calibration curves for ABZ and PRQ were prepared with in the concentration range of 4-14 μ g/ml at selected wavelengths by diluting aliquot portions of standard stock solution of each drug. The plots of Beer's law limit are shown in (Fig.4, 5, & 6)

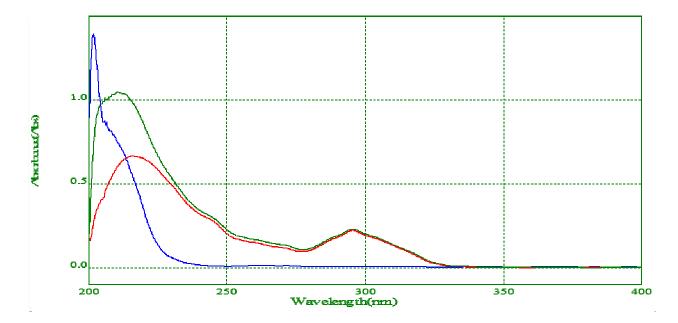
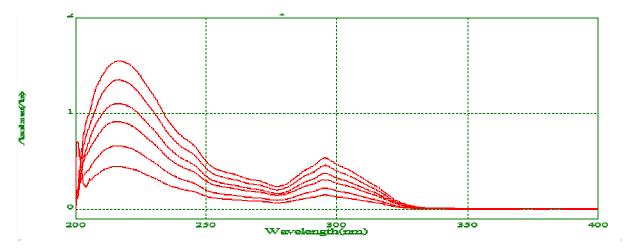


Fig: 3- Selection of detection wavelength for ABZ & PRQ at 295.4nm and 217.0nm respectively; ____: Albendazole (ABZ); ___: Praziquantel (PRQ); ___: Mixture



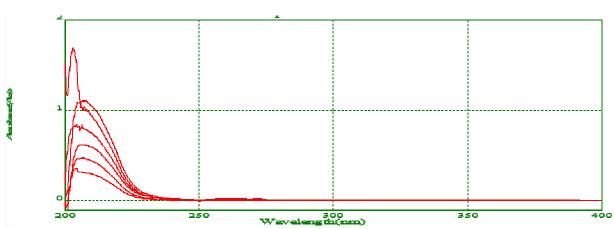


Fig: 4- Overlay Spectra of linearity of Albendazole

Fig: 5-Overlay Spectra of linearity of Praziquantel

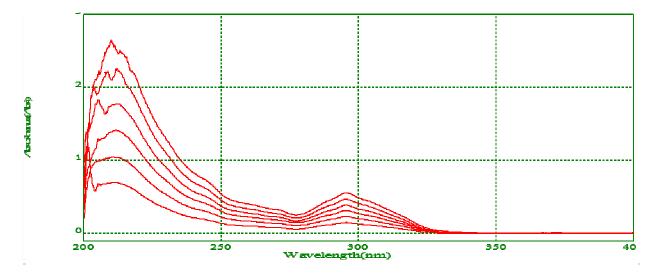


Fig: 6- Overlay Spectra of linearity of Albendazole and Praziquantel

2.4. UV Method Development

2.4.1. Preparation of Stock Solution:

Standard stock solution of Albendazole and praziquantel was prepared by dissolving individually both the drug (10mg) of in 10ml of Methanol which gives 1000μ g/ml. One ml of this stock Solution was taken and was diluted up to 10ml by using methanol (solvent) to produce a concentration of 100µg/ml solution.

2.4.2. Preparation of Working Solution:

From the above stock solution 1ml was transferred into 10ml volumetric flask and volume was made up to the mark with methanol to make 10µg/ml.

2.4.3. Determination of \lambdamax:

Then the above working sample solution $(10\mu g/ml)$ was scanned with UV-Vis Spectrophotometer in the range 400-200nm against Methanol as blank and the wavelength corresponding to absorbance was noted For Albendazole at 295.4nm and Praziquantel at 217nm for Absorbance Correction Method. (Fig-3)

2.4.4. Preparation of Calibration Curve: From the above stock solution $(100\mu g/ml)$ further dilution were made and the volume was make up to 10ml using Methanol to produce $4\mu g/ml$, $6\mu g/ml$, $8\mu g/ml$, $10\mu g/ml$ 12 $\mu g/ml$ and $14\mu g/ml$ solutions respectively. Then the construction of calibration curve was showed a straight line (Fig- 7, 8, 9, 10 & 11). The correlation coefficient was found to be 0.999 for Absorbance correction method.

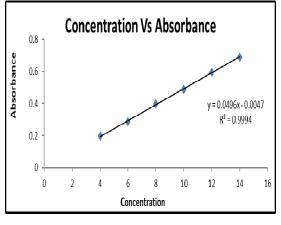


Fig: 7- Calibration Curve of "Praziquantel" at 217nm

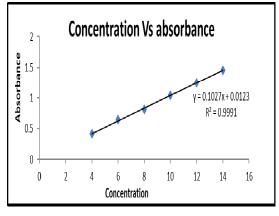


Fig: 8- Calibration Curve of "Albendazole" at 217nm

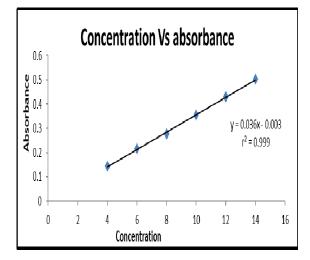


Fig: 9- Calibration Curve of "Abendazole" at 295.4nm

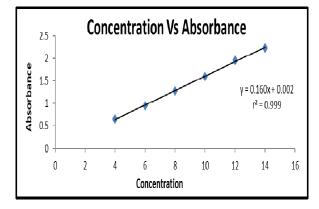


Fig: 10- Calibration Curve of "Mixture" at 217.0nm

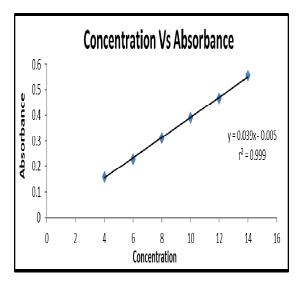


Fig: 11- Calibration Curve of "Mixture" at 295.4nm

2.5. Method Validation:

Validation is a process of establishing documented evidence, which provides a high degree of assurance that a specific activity will consistently produce a desired result or product meeting its predetermined specifications and quality characteristics.

The validation for UV method development was performed using parameters like Linearity, Accuracy, Precision, Robustness, Ruggedness, and Limit of detection (LOD), Limit of quantification (LOQ)⁽³³⁾

2.5.1. Linearity

The linearity was tested for the concentration range of 4, 6, 8, 10, 12, $14\mu g/ml$ for ABZ & PRQ. The calibration curve was constructed and evaluated by its coefficient of determination (r²). The calibration plot (concentration of PRQ

versus Absorbance of PRQ at 217nm & concentration of ABZ versus Absorbance of ABZ at 295.4nm.) was generated by replicate analysis (n =10) at all concentration levels and the linear relationship was evaluated using the least square method within Microsoft Excel® program. For minimum error with precise, concise and accurate data 10 different concentration were been taken which gave a wide range for linearity. The coefficient of determination (r^2) for ABZ & PRQ were 0.999 & 0.999 for ABZ & PRQ and given in (Fig.10, 11).

2.5.2. Accuracy:

Accuracy of the method was determined by replicate analysis of three sets of samples spiked with three different levels of ABZ & PRQ in level 80%, 100% & 120% and comparing the difference between spiked value (theoretical value) and that actually found value.

2.5.3. Precision:

The precision of the method based on within-day repeatability was determined by replicate analysis of three sets of samples spiked with three different concentrations of ABZ & PRQ (4, 10and 14μ g/ml). The reproducibility (day-to-day variation) of the method was validated using the same concentration range as described above, but only a single determination of each concentration was

made on three different days. Relative standard deviation (R.S.D.) were calculated from the ratios of standard deviation (S.D.) to the mean and expressed as percentage.

2.5.4. Specificity:

Specificity study was performed by analyzing standard solution in the presence of an excipient to find was there any interference of excipients in % recovery of ABZ & PRQ. Amount of ABZ & PRQ was spiked with 50%, 100%, and 150% of excipient (talc) and the sample was analyzed for ABZ & PRQ recovery UV-VIS Spectrophotometer.

3. Result and Discussion

3.1. Calibration Curve:

Linearity of response for ABZ and PRQ for determination of both of them in Single as well as synthetic mixture, by preparing a stock solution and suitably diluted to achieve concentration of about 4, 6, 8, 10, 12, 14µg/ml. Value of Coefficient of determination (r^2), slope and intercept were y = 0.049x - 0.004, R² = 0.999 at 217.0nm for PRQ in bulk, y = 0.102x + 0.012, R² = 0.999 at 217.0nm for ABZ in bulk, y = 0.036x - 0.003, R² = 0.999 at 295.4nm for ABZ in bulk, y = 0.160x + 0.002, $R^2 = 0.999$ at 217.0nm in synthetic mixture, y = 0.039x - 0.005, $R^2 = 0.999$ at 295.4nm in synthetic mixture. The linear regression data from the calibration curve indicate that the response is linear over the concentration range studied for both the drug. So it can be apply for determination of ABZ and PRQ in synthetic mixture.

3.2. Analytical method validation:

3.2.1. Linearity:

The linearity tested for the was concentration range of 4, 6, 8, 10, 12, 14µg/ml. and the calibration curve was constructed and evaluated by its coefficient. The correlation linear relationship was evaluated using the least square method within Microsoft Excel® program. The coefficient of determination (r^2) for both ABZ and PRQ was 0.999 given in (Table 1, 2, 3, 4, 5, & 6) and (Fig. 3, 4, & 5)

 Table No.: 1: Linearity data of Praziquantel at 217.0

Conc.	Abs.	Conc.	Conc.	Absorptivity	Avg.	
(µg/ml)	(n=5)	found	gm/100ml	gm/100ml	absorptivity	Slope:

	(nm)				gm/100ml	0.0496
4	0.1958	4.0416	0.00040	484.455		
6	0.2874	5.8884	0.00058	488.075		Intercept:
8	0.3968	8.0940	0.00080	490.235	400 011	0.0046
10	0.4882	9.9368	0.00099	491.304	489.811	
12	0.5959	12.1081	0.00121	492.146		r ² :
14	0.687	13.9448	0.00139	492.653		0.9994

Table No.: 2: Linearity data of Albendazole at 217.0nm

Conc.	Abs.	Conc.	Conc.	Absoptivity	Avg.	C alz	Slope:
(µg/ml)	(n=5)	found	gm/100ml	gm/100ml	absorptivity		0.1110
	(nm)	(µg/ml)			gm/100ml		
4	0.447	3.986	0.00039	1122.793		0.0004	Intercept
6	0.662	5.923	0.00059	1118.609		0.0005	:
8	0.912	8.172	0.00081	1116.240		0.0008	0.0051
10	1.103	9.893	0.00098	1115.154	1116.777	0.0009	
12	1.351	12.12	0.00121	1114.204		0.0012	r ² :
14	1.551	13.92	0.00139	1113.661		0.0013	0.9990

 Table No.: 3: Linearity data of Albendazole at 295.4nm

Conc.	Abs.	Conc.	Conc.	Absoptivity	Avg.	C alz	Slope:
(µg/ml)	(n=5)	found	gm/100ml	gm/100ml	absorptivity		0.039
	(nm)	(µg/ml)			gm/100ml		
4	0.1493	4.0336	0.00040	370.15		0.0003	Intercept:
6	0.2215	5.8918	0.00058	375.94		0.0005	-0.0074
8	0.3106	8.1853	0.00081	379.45		0.0008	r ² :
10	0.3754	9.8532	0.00098	380.98	378.68	0.0009	0.9990
12	0.4591	12.007	0.00120	382.33		0.0012	
14	0.5377	14.030	0.00140	383.22		0.0014	

Table No.: 4: Linearity data of Mixture at 217.0 nm

Conc.	Abs.	Conc.	Conc.	Absor-	Avg.	Alz-217	Cprq=(abs-alz	Slope:
(µg/	(n=5)	Found	gm/100	ptivity	absor-	(absorptivit	_absorptivity *	0.160
ml)	(nm)	(µg/ml)	ml		ptivity	y*Conc.)	Conc.at217)	
							/prz_absorptivi	Intercept:
							ty at 217nm	0.002
4	0.654	4.06	0.0004	1611.2		0.447	0.00042	
6	0.953	5.92	0.00059	1609.6		0.662	0.00059	r ² :
8	1.281	7.96	0.00079	1608.6		0.912	0.00075	0.999
10	1.601	9.95	0.00099	1608.1	1608.8	1.103	0.00101	
12	1.959	12.1	0.00121	1607.7		1.351	0.00124	
14	2.235	13.9	0.00139	1607.5		1.551	0.00139	

Table No.: 5: Linearity data of Mixture at 295.4 nm

Conc.	Abs.(n=5)	Conc.	Conc.	Absorptivity	Avg.	Calz	Slope:
(µg/ml)	(nm)	found	gm/100ml		absorptivity		0.160
4	0.1592	4.142	0.00041	384.30		0.00040	
6	0.2296	5.915	0.00059	388.14		0.00058	Interce
8	0.3104	7.950	0.00079	390.43		0.00079	pt:
10	0.3893	9.937	0.00099	391.76	390.10	0.00099	-0.005
12	0.4678	11.91	0.00119	392.65		0.00119	
14	0.5565	14.14	0.00141	393.35		0.00142	r ² :
							0.9990

Table No.: 6: Spectral and statistical data for determination of Albendazole andPraziquantel by Absorbance Correction Method

Parameters	Drug (n=5)	
	Albendazole	Praziquantel

295.4nm	217.0nm
y = 0.039x - 0.005	y = 0.160x + 0.002
0.2-0.9	0.2-0.5
0.9	0.5
0.2-0.7	0.2-1.7
99.79-101.95	99.09-102.19
1.32	1.31
4.0	4.0
100.74-101.24	100.15-102.49
100.176 ± 0.03	100.17 ± 0.060
	y = 0.039x - 0.005 0.2-0.9 0.9 0.2-0.7 99.79-101.95 1.32 4.0 100.74-101.24

3.2.2. Limit of detection and limit of quantification:

A limit of detection (LOD) and a limit of quantification (LOQ) were established based on the calibration curve parameters, according to the formula:

LOD = 3.3 * S.D/Slope and

LOQ = 10*S.D/Slope. (Table 7)

Table No.: 7: Determinations of LOD and LOQ.

Conc.	Abs. at	Abs. at	At 217nm	At 295.4nm
(µg/ml)	217(nm)	295.4(nm)		
4	0.6499	0.1542	<i>Avg.</i> : 0.6456	Avg.: 0.1555
4	0.6307	0.1569	S.D*: 0.0105	S.D*: 0.0024
4	0.6339	0.1565	% R.S.D: 1.6344	%R.S.D:
4	0.6505	0.1542	<i>LOD:</i> 1.31	1.51476
4	0.6539	0.1592	<i>LOQ</i> : 4.0	<i>LOD:</i> 1.32
4	0.6549	0.1525		<i>LOQ:</i> 4.0

3.2.3. Precision:

Precision was measured in terms of repeatability of measurement, performed by injecting the standard solution six times and measure the peak area. The RSD was found to be less than 2.0% for both ABZ & PRQ. This shows that Precision of the method is satisfactory which is shown in (Table 8, 9, 10.)

Intermediate Precision: Intermediate Precision with expected results and express as percentage.

Table No.: 8: Determination of Repeatability

Conc. (µg/ml)	Abs at 217.0nm	Avg.: 1.616	Abs at 295.4nm	Avg.: 0.387
10	1.605	S.D*: 0.009	0.3839	S.D*: 0.003
10	1.608	% R.S.D.: 0.5854	0.3889	% R.S.D.: 0.9012
10	1.6236		0.3833	
10	1.6299		0.3892	
10	1.614		0.3922	
10	1.6192		0.3855	

Table No.: 9: Determination of Intraday Precision
--

CONC.(µg/ml)	Abs a	t 217.0nm	Abs at 295.4nm		
4	0.3607	Avg:0.6383	0.1469	Avg:0.1475	
4	0.6339	S.D: 0.010	0.1465	S.D: 0.001	
4	0.6505	% R.S.D: 1.66	0.1492	% R.S.D: 0.98	
10	1.608	<i>Avg</i> :1.6152	0.3839	Avg:0.3853	
10	1.6236	S.D: 0.007	0.3889	S.D: 0.003	
10	0.6141	% R.S.D: 0.48	0.3833	% R.S.D: 0.79	
14	2.2356	Avg.: 2.2271	0.5565	Avg.:0.5568	
14	2.2005	<i>S.D</i> :0.023	0.5579	<i>S.D</i> :0.0009	
14	2.2453	% R.S.D: 1.05	0.5562	% R.S.D: 0.16	

Table No.: 10: Determination of Interday Precision

DAY	AT 217.0nm				AT 295.4nm			
	CONC.	Avg.	S.D*	%RSD	CONC.	Avg.	S.D*	%RSD
	(MG/ML)		(n=3)		(MG/ML)		(n=3)	
1.	4	0.7759	0.0085	1.095	4	0.2030	0.0005	0.280
	10	1.8008	0.0069	0.387	10	0.4761	0.001	0.366
	14	2.4166	0.0403	1.670	14	0.6826	0.003	0.477
2.	4	0.7769	0.0020	0.262	4	0.2111	0.001	0.521
	10	1.8510	0.0068	0.368	10	0.4774	0.001	0.346
	14	2.4672	0.0111	0.452	14	0.6728	0.002	0.311
3.	4	0.7950	0.0140	1.763	4	0.2241	0.001	0.449
	10	1.8760	0.0111	0.593	10	0.4961	0.004	0.941
	14	2.5131	0.0100	0.399	14	0.6895	0.005	0.744

3.2.4. Accuracy:

The accuracy of the method was determined by the recovery study carried out using standard addition method at three different concentration levels 80%, 100% & 120%. Resulting spiked sample solutions were assayed in triplicate and the result obtained were compare with

expected result and express as percentage. The mean % recovery of ABZ and PRQ was found to be in the range 99.79-101.95 & 98.63-101.68 % within the acceptance limit which is shown in (Table 11).

Analytes	No.	Amt.	% of	Amt.	Total	%Rec.	Mean	%RS
	of	Synthetic	Pure	Pure	Amt.		%Rec±	D
	Obs.	Mix. Add.	Drug	Drug	Found		S.D	
		(µg/ml)	Added	Added	(µg /ml)			
				(µg/ml)				
	S 1	10	80%	8	18.15	101.95	100.84	1.07
	S2	10	80%	8	17.98	99.79	±	
	S3	10	80%	8	18.06	100.79	1.08	
	S 1	10	100%	10	20.13	101.36	101.02	
ABZ	S2	10	100%	10	20.07	100.7	±	0.33
(295.4nm)	S3	10	100%	10	20.1	101	0.33	
	S 1	10	120%	12	22.18	101.5	101.0	
	S2	10	120%	12	22.08	100.69	±	0.43
	S 3	10	120%	12	22.09	100.89	0.43	
	S1	10	80%	8	17.95	98.63	100.81	1.36
	S2	10	80%	8	18.06	98.89	±	
	S3	10	80%	8	18.17	100.43	1.37	
	S 1	10	100%	10	20.12	100.18	100.34	
	S2	10	100%	10	19.98	100.1	±	0.76
	S3	10	100%	10	19.99	101.68	0.76	
PRQ	S 1	10	120%	12	21.89	99.76	100.03	
(217.0nm)	S2	10	120%	12	21.95	98.96	±	1.22
	S3	10	120%	12	22.17	101.51	1.23	

Table No.: 11: Accuracy (% Recovery Study)

3.2.5. Specificity:

Specificity study was performed by analyzing standard solution in the presence of an excipient (talc). 10mg each ABZ & PRQ were spiked with 50% (5mg), 100% (10mg), and 150% (15mg) of talc and the samples were analyzed for ABZ & PRQ recovery by UV-Spectrophotometer. Acceptance criteria for % interference <0.5%. Interference was found to be 100.74-101.24% & 100.15-102.49% for ABZ and PRQ respectively which is within the acceptance limit. Hence the excipients do not interfere with the estimation of drug which is shown in (Table 12).

Table No.: 12: Determination of % Specificity

Analyte	No. of	Excipient	Abs.	%	Avg.%	S.D*	% R.S.D
	Obs.	amount	(nm)	Recovery	Recovery		
		added					
		(mg)					
	S1	5	1.2681	99.82			
	S2	5	1.2838	101.24	100.15	0.962	0.96
	S3	5	1.2634	99.40			
Praziquantel	S1	10	1.3002	102.72			
(10mg)	S2	10	1.3034	103.00	102.49	0.660	0.64
	S3	10	1.2894	101.74			
	S1	15	1.2645	99.50			
	S2	15	1.294	102.16	100.16	1.758	1.75
	S3	15	1.2571	98.83			
	S1	5	0.3254	102.46			
	S2	5	0.3239	101.96	101.24	1.702	1.68
	S3	5	0.3159	99.3			
Albendazole	S 1	10	0.3259	102.63			
(10mg)	S2	10	0.3216	101.20	101.15	1.500	1.48
	S3	10	0.3169	99.63			
	S1	15	0.3241	102.03			
	S2	15	0.3193	100.43	100.74	1.164	1.15
	S3	15	0.3173	99.76			

3.2.6. Sample Stock Solution for Assay

Ten tablets equivalent powdered were mixed in ratio ABZ: PRQ (300mg: 25mg). A quantity of synthetic mixture powder equivalent to 65mg was taken in a 100 ml volumetric flask and diluent was added up to the mark. The solution was sonicated for 5 min. This solution was further diluted to obtain a concentration $12\mu g/ml$ for ABZ & $5\mu g/ml$ for PRQ the result is summarized in (Table 13) and UV Spectra shown in (Fig 12).

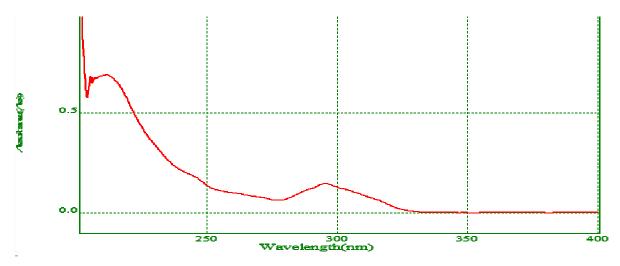


Fig: 12- Assay of synthetic mixture.

Synthetic Mixture	Drug	Label claim mg/tablet	Conc. estimated (mg)	Mean Conc. Estimated (mg)	%assay (w/w)± S.D*	% RSD
Albendazole +	Albendazole	300mg	298.79 305.59 301.24 302.75 298.03 301.24	301.27	100.42 ± 0.03	0.902
Praziquantel	Praziquantel	25mg	24.46 24.58 25.23 25.36 25.40 25.21	25.044	100.17 ± 0.067	1.639

 Table No.: 13: Determination of % Assay

4. Conclusion:

Linearity was determined at different concentration ABZ and PRQ shows linearity in the concentration range of 4-14 µg.mL-1 for ABZ and PRQ both. The percent recovery both the drugs are within the rage 95-105% which indicates the method is accurate. The % RSD values for precision are <2.0%. Method shows all validation positive response to parameter. The results of the synthetic mixture were found to be $100.42 \pm 0.03\%$ and $100.17 \pm 0.067\%$ for ABZ and PRQ respectively. The proposed Absorbance Correction method found to be simple,

rapid and sensitive. Therefore, validated UV spectrophotometric method will play a role for determination of ABZ and PRQ in their combined dosage formulation.

5. Acknowledgement:

The authors are grateful to Mercury Labs Ltd., Vadodara, India, and Micro Labs Ltd., Goa, India for providing gift samples of Albendazole and Praziquantel respectively.

References

- Campbell WC. Benzimidazoles: veterinary uses. Parasitology Today. 1990 Apr 1;6(4):130-3.
- Barragry T. Anthelmintics-review Part II. New Zealand veterinary journal. 1984 Nov 1;32(11):191-9.T.
- 3. Atkoşar Z, Altiokka G. The Determination of Albendazole by Flow Injection Analysis Method Using UV-Detection and HPLC Method in Suspensions. Journal of liquid chromatography & related technologies. 2006 Apr 1;29(6):849-56.
- Abu Zuhri AZ, Hussein AI, Musmar M, Yaish S. Adsorptive stripping voltammetric determination of albendazole at a hanging mercury drop electrode.2008 Feb,32:15, 2965-2975
- de Oliveira MF, Stradiotto NR. Voltammetric assay of albendazole in pharmaceutical dosage forms. Analytical letters. 2001 Feb 4;34(3):377-87.
- Zhao G, LI H, Liu Y, Wang Y. Application of Charge Transfer Reaction BetweenAlbendazole and Chloranilic Acid. Chinese Journal of Analytical Chemistry. 2001:400-2.
- Fregonezi-Nery MM, Baracat MM, Kedor-Hackmann ÉR, Pinheiro RM. Determination of albendazole in oral

suspension. Analytical letters. 2001 May 31;34(8):1255-63.

- Basavaiah K, Prameela HC. Titrimetric and spectrophotometric determination of albendazole with bromate and methyl orange. Oxidation Communications. 2004 Jan 1;27(1):177-85.
- 9. Saleh H, Schnekenburger J. the Colorimetric method for quantitative determination of the antibilharzial drug praziquantel and its application pharmaceutical to Analyst. preparations. 1992;117(1):87-92.
- 10. Van Tonder EC, de Villiers MM, Handford JS, Malan CE, du Preez JL. Simple, robust and accurate highperformance liquid chromatography method for the analysis of several in anthelmintics veterinary formulations. Journal of 1996 Chromatography A. Apr 5;729(1-2):267-72.
- 11. Bonato PS, De Oliveira AR, De Santana FJ, Fernandes BJ, Lanchote VL, Gonzalez AE, Garcia HH, Takayanagui OM. Simultaneous determination of albendazole metabolites, praziquantel and its metabolite in plasma by highperformance liquid chromatographyelectrospray spectrometry. mass

Journal of pharmaceutical and biomedical analysis. 2007 Jun 28;44(2):558-63.

- 12. Wu Y, Liu F, Li C. First order derivative UV-spectrophotometric determination of albendazole in tablets. Zhongguo Yiyao Gongye Zazhi. 1991;22:75-7.
- Yuan X. Derivative spectroscopic assay of praziquantel tablets. Yaowu Fenxi Zazhi. 1985;5:120-.
- 14. Saleh H, Schnekenburger J.
 Determination of praziquantel and of praziquantel in tablets by gas–liquid chromatography. Analyst. 1992 Jan 1;117(9):1457-60.
- 15. El-Khateeb SZ, El Ragehy NA, Khattab FI, Ahmad AK. Application of PMR spectrometry in quantitative analysis of praziquantel in pharmaceutical preparations. Spectroscopy letters. 1990 Apr 1;23(4):505-14.
- 16. Garcia JJ, Bolás-Fernández F, Torrado JJ. Quantitative determination of albendazole and its main metabolites in plasma. Journal of Chromatography B: Biomedical Sciences and Applications. 1999 Feb 19;723(1-2):265-71.
- 17. Morovján G, Csokán P, MakranszkiL, Abdellah-Nagy EA, Tóth K.Determination of fenbendazole,

praziquantel and pyrantel pamoate in dog plasma by high-performance liquid chromatography. Journal of chromatography A. 1998 Feb 27;797(1-2):237-44.

- 18. Chiap P, Evrard B, Bimazubute MA, De Tullio P, Hubert P, Delattre L, J. Determination of Crommen albendazole and its main metabolites ovine plasma liquid in by chromatography with dialysis as an integrated sample preparation technique. Journal of Chromatography A. 2000 Feb 18;870(1-2):121-34.
- 19. Unvermis E, Colak H, Tumer I, Ergun
 O. Study on benzimidazole anthelminthic residues in ruminant liver and kidneys by HPLC. Medycyna Weterynaryjna. 2005;61(04).
- 20. Hormazabal V, Yndestad M. Highperformance liquid chromatographic determination of praziquantel in plasma and tissues of cultured fish for residue and pharmacokinetic studies. Journal of Liquid Chromatography & Related Technologies. 1995 Feb 1;18(3):589-97.
- 21. González-Esquivel DF, Okuno CM, Rodríguez MS, Morales JS, Cook HJ. Sensitive high-performance liquid chromatographic assay for praziquantel in plasma, urine and liver

homogenates. Journal of Chromatography B: Biomedical Sciences and Applications. 1993 Mar 5;613(1):174-8.

- 22. Rogstad A, Hormazabal V, Yndestad M. Extraction of praziquantel from fish tissue and its determination by high-performance liquid chromatography. Journal of Chromatography A. 1987 Jan 1;391:328-33.
- 23. Hormazabal V, Yndestad MA. Determination of praziquantel in medicated fish feed and sediment by HPLC. Journal of Liquid & Related Chromatography Technologies. 1995 Mar 1;18(6):1231-8.
- 24. Bonato PS, Lanchote VL, Takayanagui OM. Simultaneous liquid chromatography-tandem mass spectrometric determination of albendazole sulfoxide and albendazole sulfone in plasma. Journal of Chromatography B. 2003 Jan 5;783(1):237-45.
- 25. De Ruyck H, Daeseleire E, De Ridder H, Van Renterghem R. Development and validation of a liquid chromatographic–electrospray tandem mass spectrometric multiresidue method for anthelmintics in milk.

Journal of Chromatography A. 2002 Nov 8;976(1-2):181-94.

- 26. Msagati TA, Nindi MM. Comparative study of sample preparation methods; supported liquid membrane and solid phase extraction in the determination of benzimidazole anthelmintics in biological matrices by liquid chromatography–electrospray–mass spectrometry. Talanta. 2006 Mar 15;69(1):243-50.
- 27. Procházková A, Chouki M, Theurillat R, Thormann W. Therapeutic drug monitoring of albendazole: determination of albendazole. albendazole sulfoxide, and albendazole sulfone in human plasma using nonaqueous capillary electrophoresis.

ELECTROPHORESIS:AnInternational Journal.2000Mar1;21(4):729-36.

- 28. Y. Yang, C. Ye, and G. Yao, *Zhongguo-Yiyao-Gongye-Zazhi*, infrared spectrophotometric determination of Praziquantel, 1991, 22(12), 544-5.
- 29. Pütter J. A fluorometric method for the determination of praziquantel in blood-plasma and urine. European journal of drug metabolism and pharmacokinetics. 1979 Jan 1;4(3):143-8.

- 30. Dey S, Shah S, Ghosh M, Karki N, Basak S, Sahoo NG. A Novel, Quick Column Switching RP-HPLC Guided Metabolite Profiling of Albendazole-Praziquantel in Rat Plasma: Designing New Combination Dosage Regimen with Higher Therapeutic Window. Current Analytical Chemistry. 2018 Dec 1;14(6):604-14.
- 31. Haswell SJ, Walmsley AD.
 Chemometrics: the issues of measurement and modelling.
 Analytica chimica acta. 1999 Nov 22;400(1-3):399-412.
- 32. Soto C, Contreras D, Orellana S, Yañez J, Toral MI. Simultaneous determination of albendazole and praziquantel by second derivative spectrophotometry and multivariated calibration methods in veterinary pharmaceutical formulation. Analytical Sciences. 2010 Aug 10;26(8):891-6.
- 33. ICH-Q2B Validation of Analytical Procedures, Methodology International Conference on Harmonization, 1996.